

Technical Report

Uptake of heavy metals from contaminated
soils by Salt-Marsh plants

by

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Abstract

Spartina anglica, Puccinellia maritima and Aster tripolium, three common salt-marsh species in Western Europe, were grown in contaminated sediment from the port area of Antwerp (Belgium). Growth and levels of heavy metal contamination were compared with those of Spartina alterniflora, a common salt-marsh species from the United States. The plants were grown under waterlogged and drained soil conditions. In both cases was a high and a low soil salinity maintained.

The levels of heavy metals in the shoots of the plants were generally higher under drained conditions. The difference in salinity gave no obvious differences in metal levels in the shoots. Plants grown in the same sediment, after having allowed it to dry out and get aerated, had higher levels of heavy metals in their shoots.

The different plant species showed all different levels of metals in the shoots when grown under the same circumstances: P. maritima had the lowest levels, A. tripolium the highest. In A. tripolium there was also a significant difference in metal-levels in leaf- and stem material.

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Preface

This study was conducted at the Delta Institute for Hydrobiological Research, Yerseke, The Netherlands. It started in March 1983 with Dr. A.H.L. Huiskes, dept. of Experimental Botany, and Mr. J. Nieuwenhuize, dept. of Soil Chemistry, as chief investigators.

The study is one of a number of investigations and monitoring programs of the Delta Institute for Hydrobiological Research on the dynamics of anthropogenic substances in the environment. It also links up with the experiments performed at the US Army Corps of Engineers Waterways Experiment Station, Vicksburg, USA, to develop a bioassay procedure to assess the level of contamination of dredged sediments.

The project was financed by the U.S. Army Corps of Engineers Field Verification Program, Mr. Charles C. Calhoun, Program manager. At various stages of the project assistance was received from various persons. We like to thank:

- Drs. C.R. Lee, B.L. Folsom and J.E. Simmers of WES for the numerous discussions on matters related to the project and the hospitality, they and their families extended to us during our stays in the U.S.A.
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- Mrs. M.J. van Leerdam-de Dreu, Mr. J.A. van den Ende and Mr. A.A. Bolsius for typing and for processing the text and the figures.
- Our families for allowing us to make regular journeys to the United States, for entertaining our foreign colleagues working in the same program.

1. Introduction

The boundaries between land and water have always been areas of major human activities. Large industrial complexes connected with harbour facilities are found everywhere in the world just in the interface between water and land as well as between sea and river. To keep the harbour basins open to shipping movements large quantities of sediments have to be removed regularly. These sediments are often polluted with anthropogenic substances (heavy metals, organic compounds, pesticides and oil residues) deliberately or inadvertently discharged into the water and partly adsorbed to the sediment particles. Depending on the levels of these pollutants the sediments may be a health hazard when used to create arable land or natural environments. In the United States of America one of the disposal alternatives is the creation of artificial marshes. Depending on the salinity of the dredged sediments and the disposal site these marshes may be fresh, brackish or saline.

The U.S. Army Corps of Engineers Waterways Experiment Station, Vicksburg, Mississippi is developing a bioassay for testing the quality of the dredged material, which in turn will provide evidence for the decision in what way the sediment will be disposed of.

In this study the procedure for the plant bioassay has been applied in experiments with some Dutch salt-marsh plant species growing on brackish contaminated sediment and using the North-American salt-marsh species Spartina alterniflora Loisel as a reference.

The specific objectives of the present study were:

- a. To evaluate the usefulness of the Waterways Experiment Station plant bioassay procedure by using other species and sediments;
- b. To investigate heavy metal uptake by a number of salt marsh species;
- c. To compare the heavy metal uptake of the North-American Spartina alterniflora Loisel with the uptake of native salt-marsh species, viz. Puccinellia maritima (Hudson) Parl., Aster tripolium L. and Spartina anglica Hubbard.

2. Experimental design

The design of the glasshouse experiment was according to the bioassay procedure as has been described by Folsom et al. (1981a, b).

Figure 1 shows a diagram of the experimental unit used in the experiments. A small inner bucket rested on polyvinyl chloride (PVC) pipe inside a larger outer bucket. Six 6.35 mm diameter holes were drilled in the bottom of the inner bucket. These holes were covered with a 2.54 cm thick polyurethane sponge overlaid with a 2.54 cm (approximately) layer of washed quartz sand. The sand and the sponge acted as a filter to keep the

sediment/soil from draining out the bottom of the small bucket. The holes in the small bucket allowed water movement into and out of the sediment. The water level in the inner buckets was maintained by filling the space between the outer and inner bucket up to a certain height with water.

2.1. Sediment preparation

The polluted sediment used in this study was dredged by one of the institute's research vessels R.V. "Jan Verwey" using a "Van Veen" grab. The dredging place was just outside the big shipping lock "Zandvlietsluis" in the Belgian part of the Westerschelde estuary (Fig. 2). This sediment is rather homogenous in texture and concentrations of various substances, as it is well mixed by tugs towing iron bars across the entrance of the lock in attempt to resuspend the sediment and a subsequent removal of it with the outgoing tides.

Immediately after dredging 8 samples of the sediment were taken randomly, to analyse the sediment for the various heavy metals. The dredged sediment, transported in thoroughly cleaned plastic containers, was spread out on large flats lined with PVC sheets to let it dry out. Excess water was siphoned off. After about three months, the water content of the sediment was 60.1 % on weight basis.

2.2. Plant material

The experiment was divided in two parts for reasons of available space in the greenhouse. The first year an experiment was done with Spartina alter-

niflora and Spartina anglica. Spartina alterniflora was obtained from WES transported by air. The plants were commercially grown from seeds and had 4 to 5 shoots, approximately 20 centimeters high. Spartina anglica is at present the most abundant Spartina species in the Netherlands. It was imported by the Public Works Department from Great Britain in 1924 and 1925 for land reclamation purposes and has spread since then naturally or by deliberate planting to almost all salt marshes in the Netherlands. For the experiment cuttings with 4 to 5 shoots were taken in the field and transplanted to flats filled with potting compost to allow them to root. The second year an experiment was done with S. alterniflora-cuttings and cuttings of Aster tripolium and Puccinellia maritima. The inner buckets were filled with a known amount of sediment (7000 g approximately). In the sediment the rooted cuttings were planted (or in the first experiment seedlings of S. alterniflora). The water and salinity regime in the buckets was:

- a the outer bucket filled with artificial seawater to a height just above the surface of the sediment in the inner bucket to maintain a waterlogged situation.
- b the outer bucket filled with artificial seawater of 50% of the normal strength, to a height just above the surface of the sediment in the inner bucket.
- c the outer bucket filled to a height of 5 cm with artificial seawater of normal strength, to keep a certain soil moisture by capillary rise.
- d the outer bucket filled to a height of 5 cm with artificial seawater of 50% of the normal strength.

Each treatment had 5 replicates for each species. As a reference S. alterniflora was also planted in buckets filled with a reference soil, obtained from WES. This is a loamy soil, which was mixed with a fertilizer by WES.

In the first experiment four buckets were kept waterlogged and four buckets were kept under dry conditions with artificial seawater of 50% of the normal strength. In the second experiment only two buckets with WES soil were kept as a reference under dry conditions (i.e. the outer bucket filled with 5 cm of artificial seawater of 50% of the normal strength).

In the waterlogged situation the outer buckets were regularly adjusted to the level fixed at the start of the experiment with artificial seawater. In the other buckets the level of 5 cm was kept constant too but the plants in these buckets were daily watered with demineralized water from the top. Once a week the watering was performed with artificial seawater.

2.3. Harvesting of the plants

Ninety days after the start of the experiment the shoots of the plants were cut at the soil surface with a plastic knife to avoid metal contamination. All handling of plants and sediments was done wearing plastic gloves. Almost all plants of S. anglica flowered after ninety days. The plants growing in the "upland" situation in the first experiment were harvested four weeks later, as the salinity levels in these containers were established four weeks after planting. Per treatment were for each species also one root sample and one soil sample taken from a bucket. This was only done in the first year as the buckets in the second year were kept intact for a regrowth experiment.

The shoots and roots were thoroughly rinsed with demineralized water and blotted dry with filter paper. Fresh weight of whole shoots was

measured for S. alterniflora, S. anglica and P. maritima. The shoots of A. tripolium were divided into stem, leaves and - if present - inflorescences. These fractions were separately weighted and subsequently analysed. For S. alterniflora and S. anglica also the number of tillers and the total length of the shoots were measured.

2.4. Analysis of various components in plant samples

After the measurement of the fresh weight the samples were kept at -20°C (-4°F) awaiting analyses. For the analyses the samples were freeze-dried to constant weight. The dried samples - both plant and sediment samples - were then ground in a agate mill to avoid metal contamination. All handling of samples was done, using plastic gloves. The plant samples were subsequently digested with HNO_3 conc. and H_2O_2 (30%). The destruate was brought to a volume of 50 ml by adding demineralized water. This volume contained still 2 ml HNO_3 . All elements were measured in this watery extract. Ca, Mg, Fe, Mn and Zn were measured by means of atomic absorption spectrometry, using an air-acetylene flame. Na and K were measured by means of flame - emission spectroscopy using also an air - acetylene flame.

Pb, Cd and Cu were measured on the atomic absorption spectro photometer in combination with a graphite furnace. Pb and Cd were measured by means of the standard addition method.

As was determined by hydride generation and atomic absorption spectroscopy. All measurements were performed with a Perkin Elmer model 2380 instrument.

2.5. Analysis of various components in sediment samples

The sediment samples were freeze-dried to constant weight and ground in an agate mill. The samples were sieved through a 2 mm sieve. Due care was taken to avoid heavy-metal contamination during the treatment of the samples. The chloride content of the samples was measured by potentiometric titration with AgNO_3 in a watery extract. CaCO_3 was measured by shaking 3 g of dry soil with 15 ml demineralized water and 7 ml HCl (25%). The produced CO_2 was measured volumetricly.

The particulate organic matter content was measured by pyrolysis of the sample and weighing of the produced CO_2 , adsorbed at an adsorption complex. P_2O_5 was determined colorimetrically by the vanadate method (Andersen, 1975). Total nitrogen was measured as N_2 after pyrolysis in a Carlo Erba model 1400 nitrogen analyzer. The pH was measured as pH KCl.

All metal-ions were determined as total contents after destruction of the sample with a HCl- HNO_3 mixture (3:1) and a subsequent destruction with H_2O_2 (Schramel et al., 1982). All ions were analysed in the destruate with the Atomic Absorption Spectrophotometer in combination with a graphite furnace.

3. Results

3.1. Sediment analyses

Immediately after dredging 8 soil samples were taken from the total mass of 1000 litres of sediment dredged near Antwerp (Fig. 1). The results are given in Table 1. (The individual results are given in Appendix A). Also in

Table 1 are shown the results of soil samples taken after the experiment with S. alterniflora and S. anglica in 1983. As these figures are based on just one sample no statistical treatment of the data could be performed and they have to be treated as an indication.

The figures for the moisture content of the soil and the salinity under waterlogged and drained conditions and the high and low salinity levels may indicate that the envisaged treatments were not very well maintained. Under drained conditions the salinity was low at both salinity levels, while under waterlogged conditions the salinity was comparatively high at both levels. This means that with the interpretation of the results the fact has to be kept in mind that the treatments applied were waterlogged conditions coupled with high salinity and drained conditions coupled with low salinity. Although the results of the analyses of the plants will be presented as results for the four treatments, in the discussion the data will be related to the above mentioned findings.

3.2. Plant analyses

As there are at certain points significant differences between the data of Spartina alterniflora plants grown in 1983 and plants grown in 1984, the results of the two parts of the experiment are presented separately with S. alterniflora as a reference.

3.2.1. Growth parameters

Figure 3 shows the freshweight of the shoots for S. alterniflora in the experiments of 1983 and 1984, both on dredged sediment and on WES soil.

There is a difference in average fresh weight per plant between plants grown on WES soil and plants grown on dredged sediment, this could be due to a difference in soil fertility. No significant difference could be found between the various treatments and the two years (analysis of variance). The results obtained from plants grown on WES-soil were not statistically treated, as they were too few and too different from the results on the dredged material. The individual results are given in appendix B. Fig. 4 shows the average amount of tillers produced per plant. There is a significant difference between the average number of tillers produced in 1983 and in 1984 and also a significant difference in 1983 between drained and waterlogged circumstances. The difference in tiller production could be due to the fact that in 1983 the plants were grown from seedlings and in 1984 from cuttings.

In fig. 4 the number of tillers of S. anglica, the European Spartina, is also shown, together with the LSD (= least significant difference) of the findings of both Spartina-species in 1983. The difference between the two species is obvious together with the difference between the number of tillers under drained and under waterlogged conditions. All plants of S. anglica were flowering at the end of the experiment.

In fig. 5 the average fresh weight of the two Spartina species grown in 1983 is compared. An analysis of variance showed a significant difference between the two species, but no significant difference was found due to the four treatments. Appendix C gives the individual results.

Fig. 6 shows the average fresh weights of the part of the experiment performed in 1984. The species studied were S. alterniflora, P. maritima and A. tripolium. Comparison between the results is questionable as the species are not as closely related as the Spartina's are (same genus).

Moreover A. tripolium is a forb with a completely different growth form as compared with the other two species, which are grasses. The latter two species produce a negligible amount of stem material as compared with A. tripolium. For A. tripolium stem material and leaf material are processed separately both chemically and statistically. No significant difference between the results from the various treatments could be found. There is a significant difference between the species however. This is not surprising as the stem fraction was treated statistically as a separate species. The weights of this fraction though were substantially lower as compared with the leaves fraction or the shoots of the grasses. The results of Aster tripolium show a clear difference in behaviour under the various treatments. Under low salinity conditions the plants flowered, which did not occur under high salinity conditions. Under high salinity conditions the plants form a substantial amount of dead leaves, while the amount of fresh leaves do not differ significantly. The turnover rate of leaves must therefore be higher under high salinity conditions. This has been shown for other halophytes as well (Waisel 1972). The difference between the results from high and low salinity treatments for Aster tripolium could indicate that in 1984 the two salinity levels might have been established properly, contrary to the results shown for 1983 (Table 1). This fact however could not be checked, as no soil samples were taken, for the reason that the buckets with their contents were saved for a regrowth experiment.

3.2.2. Plant analyses

Table II and Table IV show the average results of the analyses of the

shoot material of the experiments of 1983 and 1984 respectively.

The results of the experiment performed in 1983 (Table II) are the means of samples from five replicate treatments. The results shown for S. anglica grown under high salinity and waterlogged conditions are the means of three samples, as two plants showed a strongly reduced growth as compared with all other plants.

In Table III the significance of the individual figures is shown.

Although Table III shows considerable overlaps in significant differences there are some general tendencies. The influence of inundation or drainage is present for Mn, Cu, As, Na, K, Ca and Mg and to a lesser extent for Zn. Mn, Na, Ca and Mg were taken up more under inundated circumstances, while Cu, As, K and to a certain extent Zn were taken up in higher quantities under drained conditions.

Cd showed a tendency to be taken up more by S. alterniflora while Fe seemed to be taken up in higher amounts by S. anglica. The pattern for Pb is not clear. The difference in high and low salinity treatments is unclear as was already suggested in paragraph 5.1.

Table IV shows the results of the experiment performed in 1984 S. alterniflora was mainly used in this experiment to link it up with the findings in 1983.

Table V gives the significant differences between the various results. The table presents the significant differences between the species - whereby the stem - samples of A. triplium are treated as a separate species - and between the treatments, which gives a better survey in this case as compared with the method used in Table III. For most metal - levels the leaves of A. triplium score highest in a number of cases are these differences significant. The grasses have significantly lower levels of Cd in their

shoots as compared with the forb A. tripolium. This is also the case for Cu, Zn and to a lesser extent for Fe, Zn and Na. As-, K- and Ca levels show significant differences for all species. Mn- and Mg- levels are significantly lower in the stems of A. tripolium, and in P. maritima-shoots, while Pb seems to be accumulated especially in the stem of A. tripolium. No significant differences due to the treatments were found in levels of Pb, Fe, Mn and Mg. Cd, Cu and Zn were taken up more under drained circumstances. The level of As differed significantly between all treatments. Na-, K- and Cu-levels showed no obvious difference.

An analysis of variance performed on the results of S. alterniflora of 1983 and 1984 showed significant differences between the two experiments for levels of Fe, Mn, Cu, As, Na, K and Ca. The shoots of S. alterniflora had in 1984 significantly higher levels of Fe, Mn, Na, K and Ca. The levels of Cu and As were lower in 1984.

4. Conclusions and recommendations

4.1. Conclusions

The division of the total experiment into two parts, spread over two consecutive years may have obscured some of the differences in plant uptake of heavy metals and in the influence of the four treatments. Firstly the plant material of Spartina alterniflora was different in the two years. In 1983 young plants, grown from seedlings were used, in 1984 the plants were grown from cuttings taken from older plants. This may explain the difference in tiller production in 1983 and 1984. In 1983 there was a significant difference in tillers production under drained and waterlogged conditions (Fig.

4). In 1984 however no significant difference was found and the mean number of tillers was generally higher. This was also the case with the plants grown on the reference soil obtained from the Waterways Experiment Station.

Another difference between the experiment of 1983 and that of 1984 was that in 1983 sediment was used that had been dredged only two and a half months before the start of the experiment, while in 1984 the same sediment was used which by then had had about fifteen months to settle and ripen. The sediment, even over these fifteen months, never dried out; at the start of the experiment in 1984 it was still a "mousse".

These features were the reason why the experiments of the two years were discussed separately. An indication for the difference in soil conditions in 1983 and 1984 might be the higher levels of especially Mn and Fe in the shoots of S. alterniflora. It is known that a lower redox potential may result in a higher availability of these metal ions (Rozema and Roosenstein, 1985).

The differences in high and low salinity treatments were not clear. Analyses of salinity in 1983 showed that the high salinity - drained treatment had a lower salinity than the high salinity - waterlogged treatment (Table VI). This was undoubtedly due to the watering of the pots with tap water to prevent serious desiccation of the top layer of soil in the pots. This watering apparently washed down the salt, although once a week the pots were watered with artificial sea water. Although in 1984 the drained pots were watered more often with artificial sea water, the results (Table V) show that the difference in the two treatments was still not clear. As the pots were used for a sequential regrowth experiment, no soil samples were taken. This resulted effectively in three treatments: high salinity - waterlogged, low salinity - waterlogged and low salinity - drained.

The differences in heavy-metal levels in the different plant species is apparent especially in the experiment of 1984. For a bioassay procedure it is imperative to make a choice for one single-standard species. This may be Spartina alterniflora in the American situation but in the Dutch situation with in general a greater species diversity in the salt marsh it may be difficult to make a choice. Spartina anglica is a dominant species in the Dutch salt marsh and could therefore qualify. Puccinellia maritima is known as a low accumulator, while Aster tripolium is a high accumulator of heavy metals (Beeftink et al., 1982). The last species is eaten as a vegetable in the Netherlands and is in that quality subject to the standards of heavy metal contamination of the Ministry of Agriculture. It is tempting to use these standards for the bioassay with A. tripolium, to decide whether a soil is contaminated or not, but there is no jurisdiction with respect to this matter.

4.2. Recommendations

- The bioassay procedure, developed at Waterways Experiment Station, provides good method to assess the contamination level of a certain amount of dredged material.
- It has to be decided however, which species should be used as a European test organism, as different species accumulate heavy metals at different levels.
- It is clear from the results over the two years that the time between dredging and bioassay procedure is decisive for the availability of the various metal ions. It is therefore imperative to take this time into account when the results of a bioassay procedure are analysed.

5. References

- Andersen, J.M. (1975). An ignition method for determination of total phosphorus in lake sediments. *Water research* 10: 329-331.
- Beeftink, W.G., J. Nieuwenhuize, M. Stoepler and C. Mohl (1982). Heavy-metal accumulation in salt marshes from the Western and Eastern Scheldt. *The Science of the Total Environment* 25: 199-223.
- Folsom, B.L., C.R. Lee and K.M. Preston (1981a). Plant Bioassay of materials from the Blue River dredging project. Miscellaneous Paper EL-81-6, U.S. Army Engineer Waterways Experiment Station, Vicksburg, Miss.
- Folsom, B.L., C.R. Lee and D.J. Bates (1981b). Influence of disposal environment on availability and plant uptake of heavy metals in dredged material. Technical Report EL-81-12, U.S. Army Engineer Waterways Experiment Station, Vicksburg, Miss.
- Rozema, J. and J. Roosenstein (1985). Effects of zinc, copper and cadmium on the growth and mineral composition of some salt-marsh halophytes. In: Beeftink, W.G., J. Rozema and A.H.L. Huiskes (eds.), *Ecology of Coastal Vegetation*: 551-553. Junk, Dordrecht.
- Schramel, P., Li-Qiang, A. Wolff and S. Hasse (1982). ICP-Emissions-spektroskopie: Ein analytisches Verfahren zur Klärschlamm und Bodenüberwachung in der Routine. *Fresenius Z. Anal. Chem.* 313: 213-216.
- Waisel, Y. (1972). *Biology of Halophytes*. Academic Press, New York.

Figure 1. Diagram of the experimental unit, used in the experiments.

Figure 2. Map of the south-west Netherlands, with the position of the dredging area.

Figure 3. Fresh weight of shoots of Spartina alterniflora grown in 1983 and 1984 under different treatments.

Figure 4. Number of tillers of Spartina alterniflora and Spartina anglica grown in 1983 and 1984 under different treatments.

Figure 5. Fresh weight of shoots of Spartina alterniflora and Spartina anglica grown under different treatments.

Figure 6. Fresh weight of shoots of Spartina alterniflora, Puccinellia maritima and Aster tripolium grown under different treatments.

Fig. 1

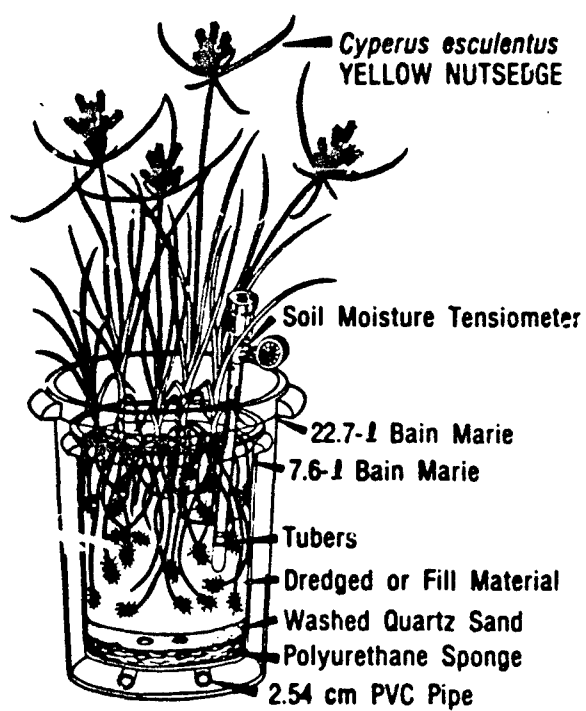
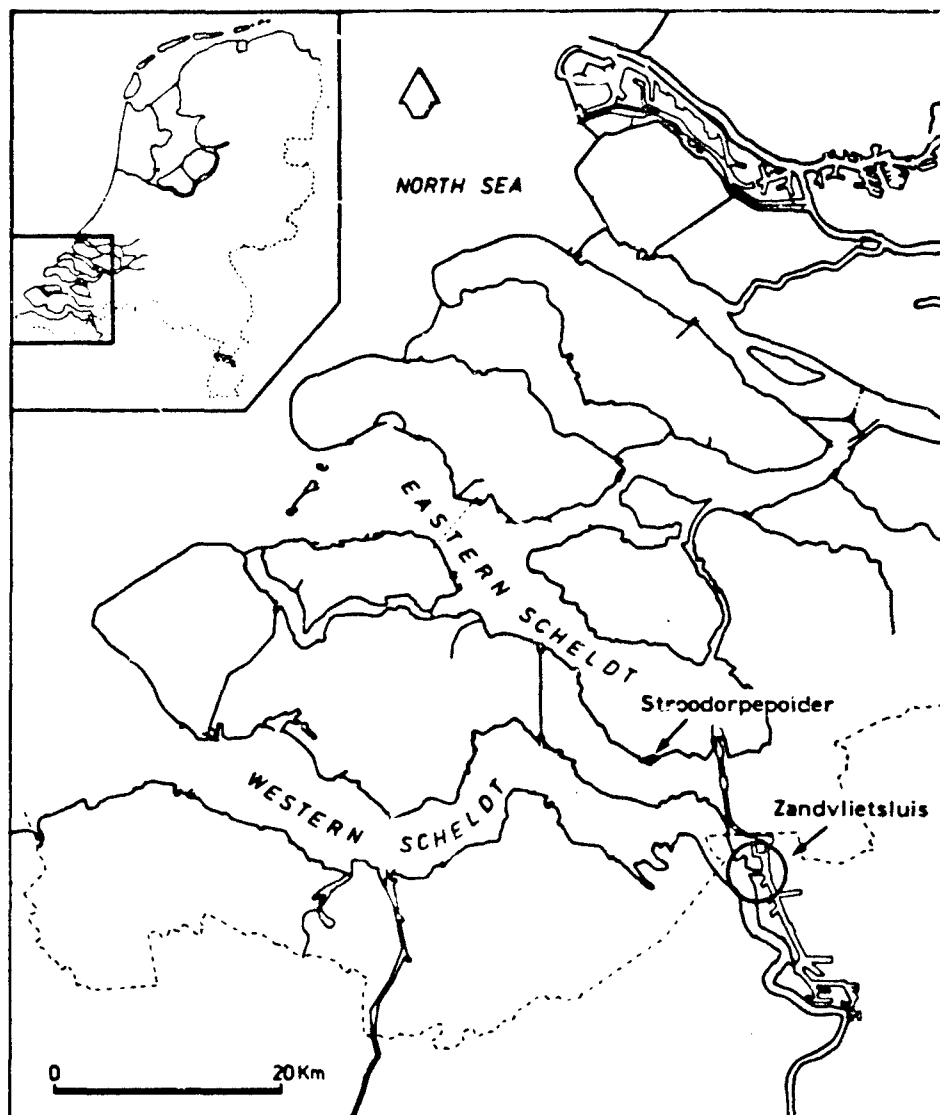
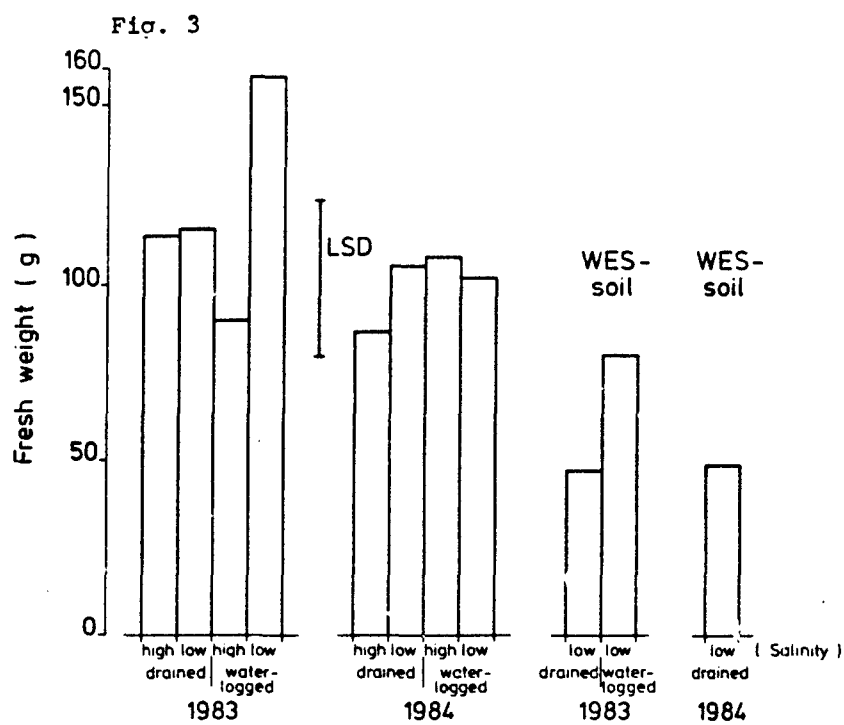


Fig. 2





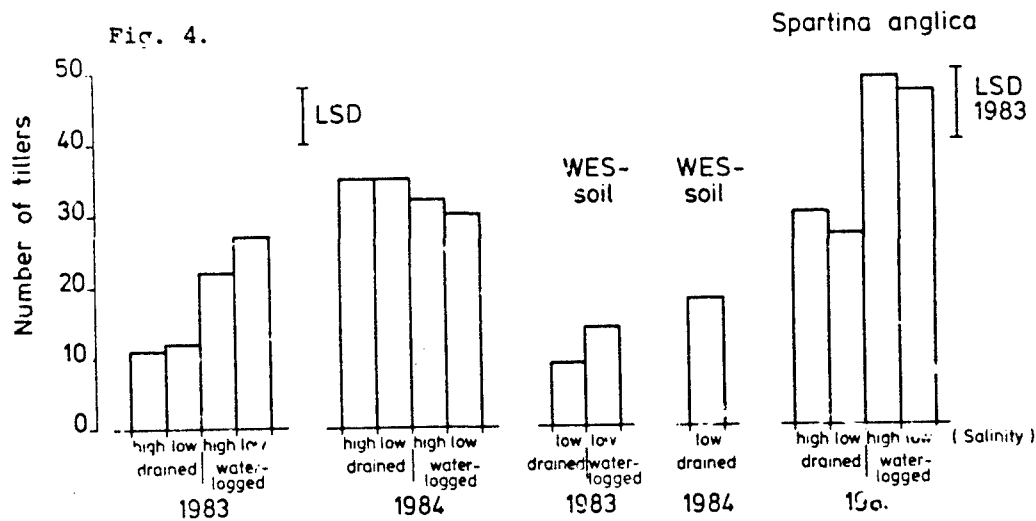


Fig. 5

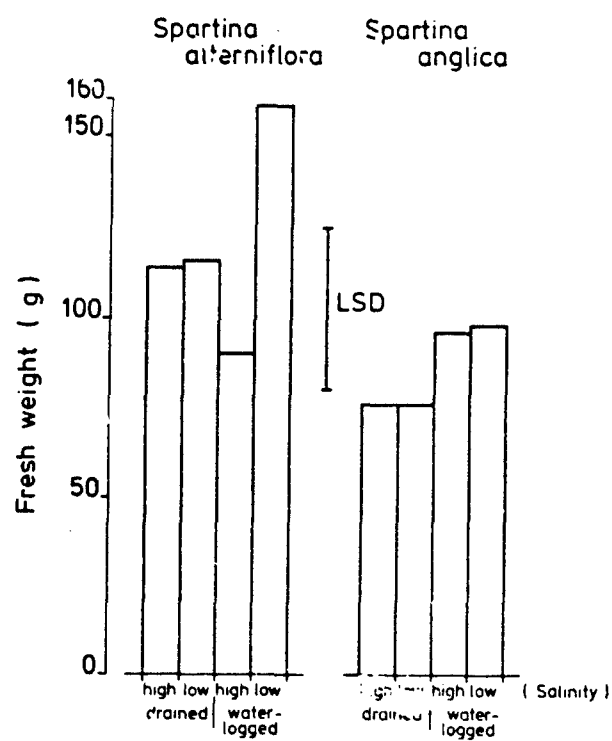
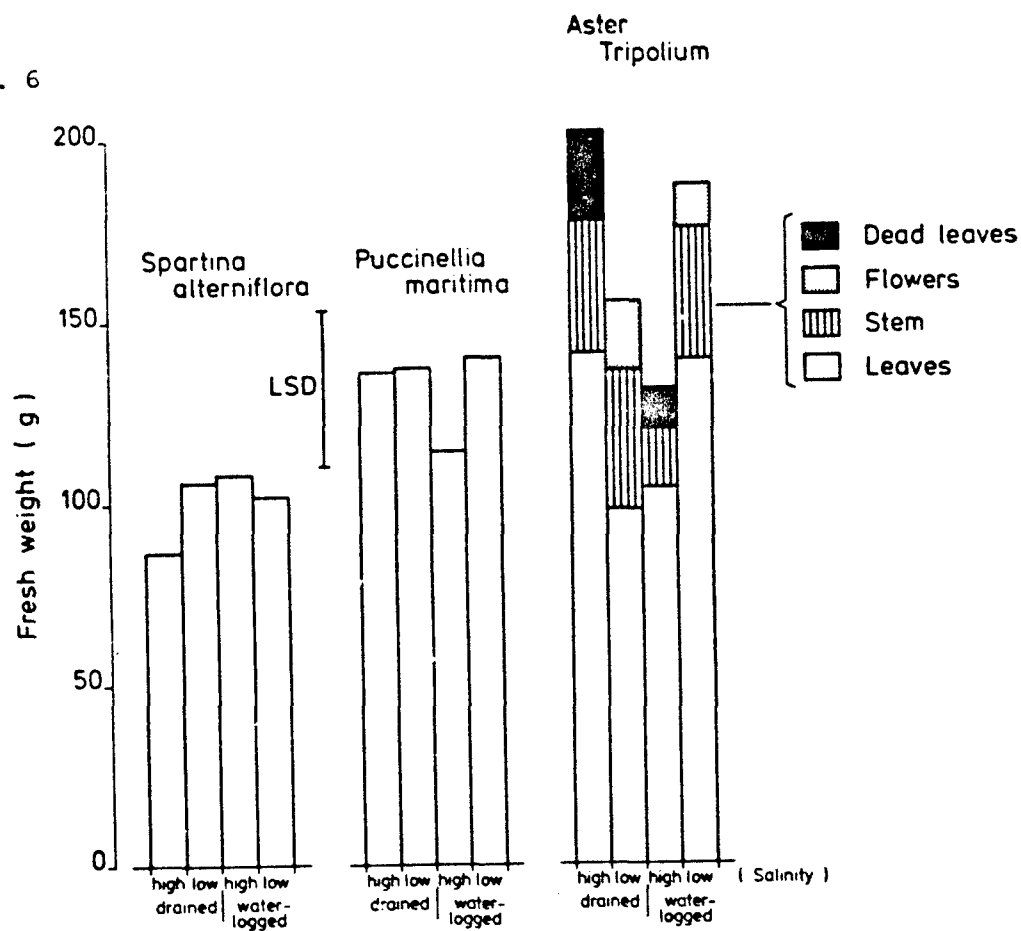


Fig. 6



le 1. Analysis results of soil samples taken immediately from the dredged sediment and taken after the sediment was left to dry out on plants were grown in it for ninety days.

	CaCO ₃	Na meq per 100 g dry soil	K meq per 100 g dry soil	moisture % per 100 g dry soil	NaCl g per 100 g dry soil	MaCl g per l water	P ₂ O ₅ mg per 100 g dry soil	pH (KCl)	POC % dry soil	H-total % dry soil	moisture % per 100 g field moist soil	clay particles < 16 µm %	Cd ppm	Pb ppm	Fe %	Cu ppm	Zn ppm	Cr ppm
8 values 8 samples a dredged trial	14.2	31.3	2.23	173.7	1.61	9.26	129	7.0	4.9	0.32	63.5	40.8	10.6	138	2.96	90	506	-
<u>rtina anglica</u> <u>/ salinity</u> erlogged	14.8			86.1	1.80	20.94	132	7.6	4.5	0.31	46.3		10.8	135	3.77	87	471	
<u>rtina anglica</u> <u>/ salinity</u> lood	13.5			66.3	0.35	5.22	125	7.7	3.6	0.27	39.9		10.7	150	3.67	88	480	
<u>rtina anglica</u> <u>/ salinity</u> erlogged	14.9			87.2	2.43	27.92	119	7.7	4.7	0.29	46.6		11.0	140	3.80	94	471	
<u>rtina anglica</u> <u>/ salinity</u> lood	14.6			87.3	0.26	2.92	110	7.7	3.8	0.29	46.6		13.6	151	4.21	101	501	
<u>rtina alterniflora</u> <u>/ salinity</u> erlogged	14.3			95.2	3.44	36.16	122	7.8	4.8	0.30	48.8		12.0	123	3.53	84	453	
<u>rtina alterniflora</u> <u>/ salinity</u> lood	13.8			76.8	0.43	5.62	119	7.6	3.6	0.27	43.4		11.7	138	3.65	97	462	
<u>rtina alterniflora</u> <u>/ salinity</u> erlogged	14.8			99.9	2.12	21.29	126	7.6	4.7	0.29	46.6		11.3	142	3.90	91	483	
<u>rtina alterniflora</u> <u>/ salinity</u> lood	13.8			62.0	0.21	3.39	128	7.5	3.5	0.26	38.3		10.7	118	3.59	81	462	
reference soil																		
ref. soil																		
<u>rtina alterniflora</u> <u>/ salinity</u> erlogged	0.65			36.5	0.04	1.15	22	6.8	0.9	0.07	26.8		1.4	112	1.23	5	37	
ref. soil																		
<u>rtina alterniflora</u> <u>/ salinity</u> lood	0.47			30.9	0.07	2.18	21	7.4	0.4	0.05	23.6		1.6	104	1.51	8	42	

Table II. Amounts of metal ions (given in mg.kg^{-1} on dry weight basis) in shoots of Spartina anglica and Spartina alterniflora grown under different salinities and soil moisture conditions.
The levels of Na, K, Cd and Mg are given in mg.g^{-1} .

	Spartina anglica		Spartina alterniflora				L.S.D.		
	high salinity		low salinity		low salinity				
	inundated	drained	inundated	drained	inundated	drained			
Cd	0.130	0.118	0.112	0.074	0.214	0.284	0.170	0.228	0.144
Pb	0.522	0.474	0.286	0.482	0.388	0.490	0.368	0.510	0.176
Fe	74.2	62.8	62.6	69.8	59.8	52.0	59.8	56.6	11.5
Mn	117.0	92.6	104.0	80.0	116.2	90.8	100.2	82.6	24.6
Cu	3.30	4.00	3.50	4.96	3.44	5.38	2.58	5.30	1.37
Zn	14.2	20.6	17.6	26.4	23.2	36.4	16.6	26.4	11.2
As	0.162	0.282	0.158	0.258	0.152	0.306	0.154	0.266	0.077
Na	33.88	23.04	31.46	20.66	24.94	16.06	22.28	14.60	3.71
K	8.52	13.86	8.92	13.62	12.62	15.98	10.18	15.20	3.05
Ca	3.10	2.32	2.78	2.08	2.84	2.54	2.70	2.36	0.52
Mg	3.70	1.94	3.54	1.82	2.18	1.86	2.34	1.72	0.51

Table III. Least significant differences (LSD) in heavy-metal concentrations in shoots of Spartina anglica and Spartina alterniflora.

The figures 1 to 8 represent the various treatments:

<u>Spartina anglica</u>	high salinity	inundated	1
		drained	2
	low salinity	inundated	3
		drained	4
<u>Spartina alterniflora</u>	high salinity	inundated	5
		drained	6
	low salinity	inundated	7
		drained	8

The underlined figures differ not significantly of each other.

Cd 4 3 2 1 7 5 8 6

Pb 3 7 5 2 4 6 8 1

Fe 6 8 5 7 3 2 4 1

Mn 4 8 6 2 7 3 5 1

Cu 7 1 5 3 2 4 8 6

Zn 1 7 3 2 5 4 8 6

As 5 7 3 1 4 8 2 6

Na 8 6 4 7 2 5 3 1

K 1 3 7 5 4 2 8 6

Ca 4 2 8 6 7 3 5 1

Mg 8 4 6 2 5 7 3 1

Table IV. Amounts of metal ions (given in $\text{mg} \cdot \text{kg}^{-1}$ on dry weight basis) in shoots of *Spartina alterniflora*, *Aster tripolium* (leaves and stems) and *Puccinellia maritima* grown under different salinities and soil moisture conditions. The levels of Na, K, Ca and Mg are given in $\text{mg} \cdot \text{g}^{-1}$.

	<i>Spartina alterniflora</i>				<i>Puccinellia maritima</i>				<i>Aster tripolium</i> (leaves)				<i>Aster tripolium</i> (stem)			
	high salinity		low salinity		high salinity		low salinity		high salinity		low salinity		high salinity		low salinity	
	inundated	drained	inundated	drained	inundated	drained	inundated	drained	inundated	drained	inundated	drained	inundated	drained	inundated	drained
Ca	0.186	0.362	0.136	0.214	0.114	0.194	0.114	0.194	0.300	0.222	0.580	0.606	1.482	0.260	0.566	0.228
Pb	0.380	0.368	0.212	0.432	0.712	0.504	0.384	0.384	1.056	0.520	0.640	0.808	0.568	1.354	0.996	0.500
Fe	253.0	153.6	176.2	160.8	75.4	92.2	143.6	143.6	79.8	259.6	208.2	168.0	481.4	523.6	405.8	185.8
Mn	126.6	100.0	104.0	115.8	69.6	68.2	61.4	61.4	71.4	81.0	130.8	141.0	168.8	29.2	40.2	29.2
Cu	3.40	3.08	3.16	3.42	4.22	5.18	3.10	3.10	3.64	5.18	7.60	5.38	6.40	13.24	19.00	10.18
Zn	20.4	21.4	20.0	23.0	11.4	13.4	10.0	10.0	13.6	29.0	37.0	45.4	79.8	23.4	34.0	20.6
As	0.200	0.144	0.122	0.194	0.230	0.228	0.160	0.160	0.334	0.096	0.030	0.112	0.070	0.298	0.242	0.136
Na	24.62	19.86	22.24	19.44	14.08	6.04	8.74	8.74	6.64	60.06	57.42	69.14	48.72	10.14	19.36	20.28
K	14.16	25.12	12.32	11.98	27.84	18.86	15.46	15.46	19.90	28.84	29.68	22.54	29.36	6.12	10.98	7.98
Cl	3.28	3.58	3.36	3.48	1.58	1.14	1.22	1.22	1.36	4.92	5.18	8.48	7.60	2.02	2.46	3.12
Mg	2.24	1.74	2.08	2.18	1.84	0.93	1.28	1.28	1.10	2.12	2.12	2.32	2.14	1.70	2.06	1.26

Table V. Least significant differences (LSD) in heavy-metal concentrations.

The figures 1 to 4 represent the plant species. The letters A to D represent the treatments Spartina alterniflora = 1; Puccinellia maritima = 2; Aster tripolium-leaves = 3; Aster tripolium-stems = 4. High salinity, inundated = C, High salinity, drained = A, Low salinity inundated = D, Low salinity, drained = B. The underlined symbols differ not significantly from each other.

	Species				Treatments			
Cd	<u>2</u>	<u>1</u>	<u>4</u>	<u>3</u>	<u>C</u>	<u>D</u>	<u>A</u>	B
Pb	<u>1</u>	<u>3</u>	<u>2</u>	4	<u>D</u>	<u>A</u>	<u>B</u>	<u>C</u>
Fe	<u>3</u>	<u>1</u>	2	4	<u>D</u>	<u>B</u>	<u>A</u>	<u>C</u>
Mn	4	2	<u>1</u>	<u>3</u>	<u>C</u>	<u>A</u>	<u>D</u>	<u>B</u>
Cu	<u>1</u>	<u>2</u>	3	4	<u>D</u>	<u>B</u>	<u>C</u>	A
Zn	<u>2</u>	<u>1</u>	4	3	<u>C</u>	<u>D</u>	<u>A</u>	B
As	3	1	4	2	D	A	B	C
Na	2	<u>1</u>	<u>4</u>	3	<u>B</u>	<u>A</u>	<u>C</u>	D
K	4	1	2	3	<u>D</u>	<u>B</u>	C	A
Ca	2	4	1	3	<u>C</u>	<u>A</u>	<u>B</u>	D
Mg	<u>2</u>	<u>4</u>	<u>1</u>	<u>3</u>	<u>B</u>	<u>A</u>	<u>D</u>	<u>C</u>

Appendix A - Complete results of the experiments

- A. Analyses of the eight soil samples taken randomly, immediately after dredging.
- B. Heavy metal concentrations and yield of Spartina alterniflora (1983).
- C. Heavy metal concentrations and yield of Spartina anglica (1983).
- D. Heavy metal concentrations and yield of Spartina alterniflora (1984).
- E. Heavy metal concentrations and yield of Puccinellia maritima (1984).
- F. Heavy metal concentrations and yield of Aster tripolium leaves (1984).
- G. Heavy metal concentrations and yield of Aster tripolium stems (1984).
- H. Quality control analysis-methods.

Appendix A. Analysis results of 8 soil samples, chosen at random, from 1000 litres contaminated material, samples with the "van Veen" grab near the Zandvlietsluis (Antwerp) - 6th January 1983. Calculated on dry basis except if mentioned

Labnr.	CaCO ₃	meq (100 g) ⁻¹ Na	meq (100 g) ⁻¹ K	moisture (100 g dry) ⁻¹ A	moisture dry g NaCl l ⁻¹ B	g NaCl (100 g) ⁻¹ C	mg (100 g) ⁻¹ P ₂ O ₅	pH	POC	Z
27096	14.1	30.4	2.35	174.4	1.73	9.91	132	6.9	4.3	0.318
27097	14.5	32.3	2.31	173.4	1.69	9.77	130	7.1	4.5	0.323
27098	13.2	29.7	2.19	174.8	1.59	9.13	126	7.0	4.7	0.321
27099	13.7	29.9	2.15	174.1	1.38	7.91	132	7.0	5.4	0.323
27100	14.6	33.2	2.26	175.5	1.64	9.35	124	7.1	4.6	0.318
27101	15.8	34.0	2.23	173.8	1.52	8.76	125	7.0	5.5	0.321
27102	14.3	30.1	2.15	171.8	1.64	9.55	133	7.0	4.9	0.329
27103	13.2	29.5	2.17	171.7	1.66	9.68	132	7.1	5.1	0.324
mean	14.2	31.3	2.23	173.7	1.61	9.26	129	7.0	4.9	0.322
S.D.	3.8	1.8	0.08	1.4	0.11	0.66	4	0.1	0.4	0.004

Labnr.	moisture (100 g wet) ⁻¹ Z	clay ≤ 16 μm Z	ppm Cd	ppm Pb	Z Fe	ppm Cu	ppm Zn	ppm Cr	C/N ratio
27096	63.6	44.9	10.1	145	2.95	86	498	197	13.5
27097	63.4	39.9	10.6	135	3.46	92	510	100	13.9
27098	63.6	40.0	10.7	136	2.76	95	498	203	14.6
27099	63.5	38.3	10.7	136	3.06	93	498	200	16.7
27100	63.7	40.5	10.8	140	3.09	88	519	201	14.5
27101	63.5	41.2	10.8	139	2.79	88	507	201	17.1
27102	63.2	40.0	10.7	137	2.46	89	510	203	14.9
27103	63.2	41.2	10.6	137	3.13	90	504	204	15.7
mean	63.5	40.8	10.6	138	2.96	90	506	201	15.1
S.D.	0.2	1.9	0.2	3	0.30	3	8	2	1.3

Appendix B. Heavy metal concentrations and yield of *Spartina alterniflora* (1983).

Pot nr.	Fresh weight shoot	nr. of tillers	ppm											mg.g ⁻¹
			Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg	
drained, high salinity														
26	118.77	10	0.37	0.48	48	113	5.1	46	0.27	17.4	16.0	2.9	1.7	
27	121.40	8	0.11	0.41	56	96	5.9	27	0.44	16.5	13.6	3.1	2.2	
28	89.60	10	0.69	0.32	54	67	6.1	65	0.14	15.9	15.1	3.0	2.1	
29	158.87	15	0.13	0.29	45	105	4.1	26	0.31	16.6	13.0	2.2	1.9	
30	81.31	10	0.12	0.95	57	73	5.7	18	0.37	13.9	22.2	1.5	1.4	
drained, low salinity														
36	132.06	13	0.24	0.53	50	82	5.1	30	0.23	16.7	14.1	2.3	1.7	
37	112.48	14	0.15	0.44	66	56	5.7	18	0.31	14.5	16.7	1.9	1.4	
38	91.52	11	0.19	0.60	62	67	5.8	20	0.31	12.8	15.9	2.6	1.9	
39	152.72	14	0.42	0.51	44	130	5.4	41	0.22	13.1	15.7	3.3	2.2	
40	93.69	9	0.14	0.47	61	78	4.5	23	0.26	15.9	13.6	1.7	1.4	
inundated, high salinity														
21	85.50	15	0.17	0.57	64	74	4.6	30	0.21	21.6	10.7	2.7	2.7	
22	131.91	22	0.32	0.29	67	133	5.2	27	0.18	24.5	13.0	2.9	2.0	
23	55.84	10	0.17	0.30	61	153	2.9	22	0.13	33.8	13.2	3.1	3.0	
24	139.86	33	0.29	0.43	55	122	2.4	18	0.14	24.7	12.7	2.7	1.6	
25	38.99	29	0.12	0.35	52	99	2.1	19	0.10	20.1	13.5	2.8	1.6	
inundated, low salinity														
31	155.28	23	0.09	0.32	63	109	2.2	16	0.15	24.3	9.5	2.9	2.3	
32	99.90	27	0.19	0.29	71	105	2.5	14	0.16	23.1	9.5	3.0	2.3	
33	119.32	24	0.15	0.49	66	83	2.2	17	0.16	22.6	11.3	2.2	1.6	
34	158.11	27	0.19	0.48	63	108	2.9	17	0.17	22.6	9.5	3.0	2.4	
35	258.83	32	0.23	0.26	46	96	3.1	19	0.13	19.8	11.1	2.4	2.1	

Appendix. C. Heavy metal concentrations and yield of *Spartina anglica* (1983).

Pot nr.	Fresh weight shoot	nr. of tillers	ppm										mg.g ⁻¹			
			Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg			
drained, high salinity																
6	73.66	25	0.09	0.59	77	86	4.9	24	0.30	24.5	13.4	1.9	1.7			
7	87.38	24	0.29	0.46	56	122	2.4	19	0.15	27.6	16.0	2.9	2.2			
8	73.30	23	0.07	0.30	67	89	3.6	18	0.35	18.9	13.9	2.0	1.6			
9	87.91	53	0.03	0.46	47	90	2.6	12	0.32	23.2	7.8	2.7	2.4			
10	59.34	24	0.11	0.56	67	76	6.5	30	0.29	21.0	18.2	2.1	1.8			
drained, low salinity																
16	116.30	34	0.05	0.41	59	64	3.5	20	0.18	20.0	8.8	2.1	1.0			
17	56.14	24	0.06	0.63	69	79	5.9	31	0.25	21.5	17.0	2.3	1.8			
18	79.50	20	0.11	0.33	72	83	7.2	36	0.23	24.5	15.9	2.0	2.0			
19	62.68	26	0.05	0.35	83	85	3.7	18	0.34	18.6	13.4	1.7	1.6			
20	63.11	31	0.10	0.69	66	89	4.5	27	0.29	18.7	13.0	2.3	1.8			
inundated, high salinity																
1	18.14	18	0.29	1.4	229	91	6.4	32	0.30	39.8	10.2	4.2	4.8			
2	3.12	2	0.23	1.2	244	64	8.9	21	-	43.9	10.4	4.1	5.5			
3	93.83	51	0.14	0.49	69	122	3.5	12	0.16	34.6	8.8	3.6	3.6			
4	63.24	34	0.13	0.50	75	127	3.6	18	0.17	37.7	9.4	3.2	4.3			
5	131.82	62	0.12	0.58	79	102	2.8	13	0.16	29.3	7.4	3.1	3.2			
inundated, low salinity																
11	127.17	57	0.11	0.21	46	112	3.3	14	0.14	33.1	8.7	3.0	3.5			
12	105.82	58	0.04	0.21	53	90	2.8	15	0.11	31.7	7.7	2.6	3.6			
13	120.06	47	0.14	0.40	74	98	3.0	17	0.19	28.7	7.9	2.7	4.0			
14	46.19	27	0.16	0.26	75	122	4.6	26	0.22	32.7	11.1	2.8	3.4			
15	89.96	46	0.11	0.35	65	98	3.8	16	0.13	31.1	9.2	2.8	3.2			

Appendix. D. Heavy metal concentrations and yield of *Spartina alterniflora* (1984).

Pot nr.	Fresh weight shoot	nr. of tillers	ppm										mg.g ⁻¹			
			Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg			
drained, high salinity																
41	125.47	33	0.42	0.46	183	116	3.4	17	0.17	21.5	23.5	3.4	1.9			
42	66.65	26	0.28	0.30	131	78	2.8	17	0.11	18.5	24.3	3.1	1.5			
43	84.34	37	0.40	0.54	158	101	3.3	25	0.15	18.0	27.0	3.5	1.8			
44	92.59	43	0.39	0.36	159	107	3.4	28	0.12	22.0	26.8	4.0	1.9			
45	67.84	37	0.32	0.18	137	98	2.5	20	0.17	19.3	24.0	3.9	1.6			
drained, low salinity																
46	99.69	37	0.35	0.42	189	124	3.6	28	0.22	17.3	11.9	3.9	2.0			
47	107.30	40	0.22	0.36	137	116	4.2	23	0.17	22.4	12.7	4.3	2.5			
48	84.40	37	0.35	0.54	161	94	3.0	25	0.20	17.0	14.6	3.2	1.6			
49	100.90	31	0.12	0.64	226	101	3.0	23	0.25	22.3	12.6	2.9	2.2			
50	136.12	32	0.03	0.20	91	144	3.3	16	0.13	18.2	8.1	3.1	2.6			
inundated, high salinity																
51	107.83	39	0.19	0.16	104	124	3.7	22	0.11	26.5	22.0	3.7	2.8			
52	126.88	44	0.26	0.20	172	114	3.1	23	0.12	20.8	10.8	2.8	2.0			
53	91.47	22	0.17	0.30	203	139	3.3	21	0.14	24.0	13.0	3.1	2.0			
54	94.24	23	0.12	0.38	282	122	3.4	18	0.26	27.8	13.5	3.4	2.2			
55	119.73	32	0.19	0.86	504	134	3.5	18	0.37	24.0	11.5	3.4	2.2			
inundated, low salinity																
56	58.29	21	0.11	0.40	128	106	3.6	24	0.12	26.3	15.6	3.9	2.5			
57	112.41	26	0.11	0.20	365	103	3.0	16	0.12	21.4	9.7	3.3	2.1			
58	124.77	39	0.20	0.24	167	143	2.9	19	0.11	21.0	13.1	3.5	1.9			
59	95.57	36	0.13	0.10	117	91	3.0	20	0.11	21.2	12.5	3.0	1.9			
60	120.58	28	0.23	0.12	104	87	3.3	21	0.15	21.3	10.7	3.1	2.0			

Appendix. E. Heavy metal concentrations and yield of *Puccinellia maritima* (1984).

Pot nr.	Fresh weight shoot	ppm										mg.g ⁻¹			
		Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg			
drained, high salinity															
21	130.21	0.17	0.58	210	62	4.4	14	0.29	4.8	16.0	0.9	0.8			
22	134.87	0.35	1.20	450	79	5.7	16	0.43	6.8	17.3	1.5	1.1			
23	127.23	0.13	0.16	117	65	5.3	11	0.11	7.3	21.5	1.3	1.1			
24	141.88	0.13	0.24	88	66	6.2	13	0.12	5.5	19.5	0.9	0.8			
25	147.49	0.19	0.34	176	69	4.3	13	0.19	5.8	20.0	1.1	0.9			
drained, low salinity															
26	141.02	0.25	1.32	829	77	3.5	15	0.57	6.3	20.6	1.3	1.0			
27	127.32	0.39	2.54	1000	78	4.3	19	0.75	11.0	22.6	2.3	1.8			
28	124.00	0.29	0.46	204	77	3.9	12	0.18	5.8	22.1	1.1	0.9			
29	155.77	0.28	0.66	259	54	3.7	12	0.15	5.3	16.6	1.1	0.9			
30	137.00	0.29	0.30	115	71	2.8	10	0.12	4.8	17.6	1.0	0.9			
inundated, high salinity															
31	97.27	0.06	0.14	97	63	3.6	11	0.07	9.0	16.3	1.2	1.5			
32	132.59	0.19	1.06	343	75	5.2	14	0.31	13.3	18.8	1.9	1.8			
33	109.13	0.10	0.54	126	54	3.8	12	0.14	16.0	17.0	1.6	2.4			
34	124.62	0.09	0.86	425	64	4.0	10	0.39	11.3	38.3	1.4	1.4			
35	104.78	0.13	0.96	306	92	4.5	10	0.24	20.8	48.8	1.8	2.1			
inundated, low salinity															
36	122.93	0.24	0.60	230	64	2.9	11	0.21	8.4	15.6	1.3	1.3			
37	116.24	0.05	0.34	131	56	2.6	10	0.18	8.5	18.1	1.2	1.3			
38	175.10	0.19	0.52	149	57	2.7	11	0.20	9.0	13.6	1.3	1.4			
39	158.79	0.06	0.26	226	65	4.2	9	0.14	9.9	15.8	1.3	1.3			
40	127.23	0.03	0.20	104	65	3.1	9	0.07	7.9	14.2	1.0	1.1			

Appendix. F. Heavy metal concentrations and yield of Aster tripolium leaves (1984).

Pot nr.	Fresh weight shoot	mg. g ⁻¹										
		Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg
drained, high salinity												
1	144.84	0.59	0.54	95	159	7.7	46	0.04	72.3	36.6	5.2	2.4
2	168.03	0.78	0.76	60	120	5.3	45	0.01	46.4	24.8	4.1	2.0
3	0.0	-	-	-	-	-	-	-	-	-	-	-
4	156.18	0.51	0.84	153	144	7.5	27	0.05	65.0	28.3	6.2	2.3
5	96.67	0.44	0.42	61	120	9.9	30	0.02	46.0	29.0	5.2	1.8
drained, low salinity												
6	51.68	0.83	0.48	118	181	5.4	98	0.15	53.0	25.1	6.6	2.1
7	161.13	0.52	0.36	72	87	5.2	31	0.04	45.2	26.1	5.5	1.8
8	156.93	0.59	0.60	68	70	6.6	29	0.04	43.7	44.4	5.1	2.2
9	57.36	3.14	0.70	72	281	5.1	138	0.06	52.5	24.6	11.2	2.4
10	61.00	2.33	0.70	69	225	6.7	103	0.06	49.2	26.6	9.6	2.2
inundated, high salinity												
11	95.76	0.17	0.54	55	81	5.5	33	0.04	58.8	23.8	4.9	1.3
12	71.79	0.37	0.72	67	40	3.6	24	0.10	45.0	35.5	3.6	4.6
13	93.18	0.24	0.52	81	106	5.4	32	0.12	60.5	25.5	5.9	1.7
14	154.00	0.17	0.32	96	102	5.8	25	0.09	63.6	27.0	4.8	1.4
15	105.28	0.16	0.50	78	76	5.6	31	0.13	72.4	32.4	5.4	1.6
inundated, low salinity												
16	90.69	0.26	0.52	180	163	5.6	45	0.11	58.0	14.8	8.4	1.9
17	157.82	0.18	0.52	58	65	5.6	18	0.01	65.0	25.1	6.6	2.1
18	218.58	0.36	0.80	48	112	6.8	22	0.05	80.6	23.3	8.1	2.5
19	212.38	0.19	0.44	57	71	3.4	22	0.03	54.2	26.9	5.1	1.5
20	15.26	2.04	1.76	375	294	5.5	120	0.36	87.9	22.6	14.2	3.6

Appendix. G. Heavy metal concentrations and yield of Aster tripolium stems (1984).

Pot nr.	Fresh weight stem	ppm										mg.g ⁻¹			
		Cd	Pb	Fe	Mn	Cu	Zn	As	Na	K	Ca	Mg			
drained, high salinity															
1	103.68	0.27	0.12	152	31	7.6	17	0.13	17.6	13.9	1.4	0.9			
2	27.12	0.43	0.46	124	21	16.3	27	0.05	6.4	4.3	1.2	1.5			
3	6.53	1.10	2.38	842	75	29.2	68	-	54.5	23.0	5.4	5.5			
4	18.24	0.54	1.00	477	36	19.0	23	0.33	7.5	4.0	1.8	1.0			
5	23.19	0.49	1.02	434	38	22.9	35	0.46	10.8	9.8	2.5	1.4			
drained, low salinity															
6	77.67	0.63	0.60	146	35	4.7	25	0.20	32.4	17.6	2.6	1.2			
7	8.99	0.59	0.70	94	20	17.4	21	0.12	9.3	4.3	2.1	1.3			
8	3.68	0.59	0.70	94	20	17.4	21	0.12	9.3	4.3	2.1	1.3			
9	52.24	1.57	0.14	50	42	6.7	30	0.08	15.6	15.1	3.5	1.3			
10	48.76	1.14	0.20	42	37	8.2	28	0.07	18.1	18.1	3.4	1.2			
inundated, high salinity															
11	21.56	0.09	0.22	146	14	10.9	16	0.12	8.8	7.8	1.4	1.4			
12	5.24	0.68	4.06	1561	66	14.6	31	-	11.0	6.8	3.0	2.1			
13	18.38	0.16	0.86	149	16	14.5	21	0.21	8.0	4.0	1.5	1.3			
14	21.23	0.14	0.83	397	21	13.2	18	0.35	6.4	6.8	1.6	1.4			
15	17.79	0.23	0.80	365	29	13.0	31	0.51	16.5	5.2	2.6	2.3			
inundated, low salinity															
16	85.90	0.18	0.48	218	62	5.1	21	0.11	31.6	12.6	5.4	1.2			
17	14.71	0.17	0.64	301	21	13.4	17	0.16	13.6	4.0	2.4	1.0			
18	23.37	0.22	0.50	142	23	12.8	19	-	16.3	4.3	2.7	1.2			
19	18.55	0.16	0.32	120	17	11.8	15	0.13	9.8	3.4	2.3	1.3			
20	38.14	0.41	0.56	148	33	7.8	31	0.14	30.1	15.6	2.8	1.6			

Appendix H.

Quality control (plant samples)

	Fe	Zn	Mn	Cu	Pb	Cd	Cr	Ca(%)	K(%)	Mg(%)
752	64	237	9.5	6.5	3.1	5.0	2.50	4.56	0.68	0.68
705	57	228	9.3	5.5	3.4	5.0	2.93	4.21	0.63	0.63
700	60	244	9.9	5.6	3.5	4.9	3.11	4.58	0.68	0.68
739	62	238	10.3	5.6	3.7	4.9	3.03	4.30	0.66	0.66
722	61	239	9.4	5.2	3.5	5.2	2.95	4.46	0.66	0.66
745	59	244	9.9	6.3	3.2	5.0	3.00	4.51	0.68	0.68
700	61	244	9.8	6.3	3.3	5.0	3.06	4.46	0.68	0.68
723	60	239	9.7	5.9	3.4	5.0	3.00	4.44	0.67	0.67
standard deviation	22.2	2.22	5.78	0.351	0.471	0.177	0.105	0.07	0.137	0.019
standard error	8.40	0.84	2.19	0.133	0.178	0.067	0.040	0.028	0.052	0.0071
coefficient of variation	3.1	3.7	2.4	3.6	8.1	5.2	2.1	2.5	3.1	2.8
95% confidence limits	20	2	5	0.35	0.45	0.15	0.10	0.07	0.13	0.01
limits of detection	0.45	0.05	0.04	0.57	2.2	0.06	2.8	0.03	0.01	0.01
limit *	690±25	62±6	238±7	11±1	6.3±0.3	(3) 4.5±0.5	3.00±0.03	4.46±0.03	(0.7)	(0.7)
NBS 1573 (certified values)										

* standard deviation x 3 (n=6) Fe, Zn, Mn, Ca, K, Mg in ppm en Cu, pb, Cd, Cr in ppb
() = non-certified values, information only

Samples are analysed according to the analysis methods of the soil science laboratory DIHO, 1984.
15 February 1985 - Soil Science Laboratory